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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds
SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

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Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium

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phosphate SnS 2 d. tin(IV) sulfide HNO 2
e. nitrous acid 2.

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CHAPTER 5 REVIEW The Periodic Law
SECTION 1 SHORT ANSWER Answer the following questions in the space provided.

1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

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